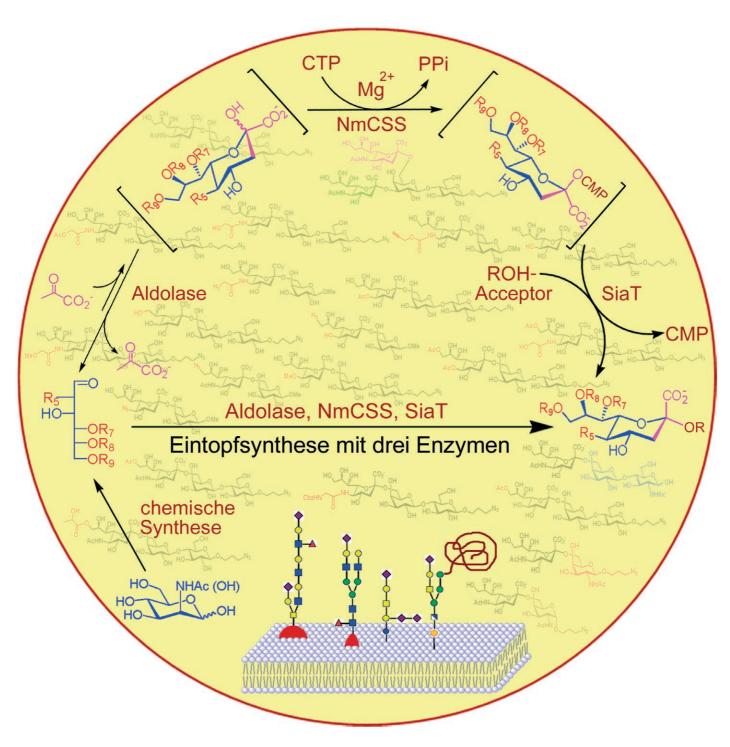


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 $\alpha\text{-}2,6\text{-}Verknüpfte Sialoside, die sowohl natürliche als auch nichtnatürliche Sialinsäuren enthalten, wurden in einer chemoenzymatischen Eintopfsynthese mit drei Enzymen erhalten. In der Zuschrift auf den folgenden Seiten zeigen X. Chen et al., dass die relaxierte Substratspezifität der Biosynthese-Enzyme der Schlüssel ist.$



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Enzyme Catalysis

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Highly Efficient Chemoenzymatic Synthesis of Naturally Occurring and Non-Natural α-2,6-Linked Sialosides: A P. damsela α-2,6-**Sialyltransferase with Extremely Flexible Donor-Substrate Specificity****

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Sialic acids are a family of α -keto acids with a nine-carbon backbone. They have been predominantly found as terminal carbohydrate units on glycoproteins and glycolipids of vertebrates or as components of capsular polysaccharides and lipooligosaccharides of pathogenic bacteria.^[1] Sialic acid containing structures play pivotal roles in many physiologically and pathologically important processes, including cellular recognition and communication, bacterial and viral infection, and tumor metastasis, etc.[1] Currently, more than 50 structurally distinct forms of sialic acids have been found in nature,[1] more than 15 of which have been found on human red-blood-cell (RBC) surfaces, saliva proteins, and gastrointestinal mucins.^[2] Three basic forms of sialic acids (Figure 1)

Figure 1. Three basic forms of naturally occurring sialic acids. Ac = acetyl.

are N-acetylneuraminic acid (Neu5Ac), N-glycolylneuraminic acid (Neu5Gc), and deaminoneuramine acid (KDN). Based on these three forms, single or multiple substitutions can occur at the hydroxyl group on C4, C5, C7, C8, and/or C9 positions, including O-acetylation and the less frequent Omethylation, O-lactylation, O-sulfation, and O-phosphorylation (Figure 2).^[1]

Modifications of sialic acids and cell-surface presentation of modified sialic acids are species and tissue specific. They

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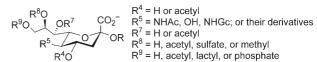


Figure 2. Naturally occurring sialic acid modifications.

are developmentally regulated and are believed to be closely related to their biological functions.^[1] Nevertheless, a clear understanding of the mechanism and the significance of nature's sialic acid structural diversity is currently missing. This is mainly owing to the difficulties in obtaining homogenous sialosides or sialylglycoconjugates, especially those that contain diverse naturally occurring sialic acid modifications. These structures are extremely difficult to isolate in homogenous forms from natural sources^[3] and chemical sialylation remains challenging.^[4] Although sialyltransferase-catalyzed synthesis offers great advantages,^[5] it suffers from the low expression level and the narrow substrate specificity of many sialyltransferases, especially those from mammalian sources. [6] Current chemical^[4,7] and enzymatic^[5,8] sialylation activities have been focusing on structures containing non-natural sialic acid derivatives and a limited number of natural sialic acid forms (e.g. Neu5Ac, Neu5Gc, KDN, Neu5Ac8Me, and Neu5,9Ac₂).^[3-9] Further development of the synthetic methodology, thus, is required to obtain structurally defined naturally occurring sialosides for better understanding of their biological roles.

Sialic acid modifications, such as O-acetylation, O-methylation, O-lactylation, and O-sulfation, are believed to occur after the formation of sialylglycoconjugates or polysialic acids in mammals, [10] group C meningococci, [11] and E. coli. [12] There is indication, however, that O-acetylation may occur on free Neu5 Ac in the biosynthesis of Group B Streptococcus (GBS) capsular polysaccharide. [10b] Only a few enzymes involved in the sialic acid modifications have been discovered, including 9-O-acetyltransferases from rat liver, [10a] bovine submandibular gland, [13] E. coli K1, [15,14] and C. jejuni; [15] 4-O-acetyltransferases from guinea-pig liver^[16] and equine submandibular gland: [17] a sialic acid 8-O-methyltransferase from starfish A. rubens. [18] Cytidine monophosphate (CMP)-Neu5 Ac hydroxylase^[19] and *C. jejuni* 9-*O*-acetyltransferase^[15] are the only proteins in this category that have ever been cloned. Owing to the unavailability of these sialic acid modifying enzymes, it is impractical to synthesize naturally occurring sialosides by totally following their biosynthetic pathways.

Instead of nature's way of introducing sialic acid modifications after oligosaccharide formation, we have established a highly efficient and convenient one-pot, three-enzyme chemoenzymatic approach for the synthesis of sialosides containing naturally occurring as well as non-natural sialic acid modifications at C5, C7, C8, and/or C9. In this method, sialic acid modifications can be chemically introduced at the very beginning, onto the six-carbon sugar precursors (ManNAc or mannose) of sialic acids. These ManNAc or mannose analogues can then be directly converted to the corresponding sialosides in one pot by using three enzymes, including a sialic acid aldolase, a CMP-sialic acid synthetase, and a sialyltransferase without the isolation of intermediates

(Scheme 1). Depending on the type of the sialyltransferase used, α -2,3- or α -2,6-linked sialosides can be synthesized conveniently. Combining the diversity of organic synthesis and the highly efficient, regio- and stereoselective enzymatic approaches, this chemoenzymatic system is an attractive synthetic method for complex sialic acid containing structures

Two major challenges for highly effective chemoenzymatic synthesis are 1) the availability and efficiency of biosynthetic enzymes; and 2) the extent of substrate modification that can be tolerated by the enzymes. We have successfully found solutions for synthesizing complex sialosides by matching these two challenges.

Previously, we reported the high-level expression of a recombinant sialic acid aldolase from $E.\ coli\ K-12$ and a CMP-sialic acid synthetase from $N.\ meningitidis\ (NmCSS).^{[20]}$ Both enzymes have flexible substrate specificity. They have been used successfully in one-pot, two-enzyme preparative syntheses of CMP-sialic acid derivatives[^{20]} and one-pot, three-enzyme syntheses of α -2,3-linked sialoside libraries. To prepare α -2,6-linked sialosides with naturally occurring or non-natural sialic acid modifications, we investigated the donor–substrate specificity of a recombinant $Photobacterium\ damsela\ \alpha$ -2,6-sialyltransferase (Pd2,6ST) and explored its application in the one-pot, three-enzyme synthesis.

Pd2,6ST was the first bacterial α -2,6-sialyltransferase that has ever been cloned. [22] The isolation of this sialyltransferase was firstly reported in 1996. [23] Subsequently, its extremely flexible acceptor–substrate specificity was described. [8c,24] No study, however, has been reported on the donor–substrate specificity of this enzyme.

To facilitate the protein purification, a truncated Pd2,6ST encoding for amino acid residues 16–497 of the full-length protein was cloned into a pET15b vector and expressed as a N-His₆-tagged protein. The recombinant Pd2,6ST was purified by a nickel-nitrilotriacetic acid (Ni-NTA) agarose affinity column. The tolerance of donor–substrate modification of the purified Pd2,6ST was firstly tested by TLC analysis (EtOAc/MeOH/H₂O/HOAc = 4:2:1:0.1) in the one-pot, three-enzyme system shown in Scheme 1, in which CMP-sialic acid derivatives were generated in situ from sialic acid precursors catalyzed by the aldolase and NmCSS. For sialosides that do not contain O-acetyl or O-lactyl groups, reactions were typically performed in a tris(hydroxymethyl)aminomethane

(Tris)-Cl buffer solution at pH 8.5 for 2–10 h at 25 °C. When this condition was used for the synthesis of sialosides containing O-acetyl groups, de-O-acetylation was observed. Therefore, a Tris-Cl buffer solution (pH 7.5) with a lower pH value was used for synthesizing sialosides containing O-acetylated or O-lactylated sialic acid residues to avoid the hydrolysis of the esters. After observing the product formation on TLC, preparative scale syntheses were carried out. Sialoside products were purified by gel filtration chromatography and characterized by NMR spectroscopy and HRMS.

As shown in Table 1, by using 3-azidopropyl lactoside **20** as an acceptor for Pd2,6ST, naturally occurring α-2,6-linked sialosides containing Neu5Ac or its C5-substituted analogues (**26–32**) were firstly synthesized in 75–99% yields (Table 1, entries a–g) from ManNAc, mannose, or their C2-modified analogues (**1–7**) as sialic acid precursors. 5-*O*-Acetyl-KDN (KDN5Ac) (the sialic acid form in **29**) has been found on the eggs of *Rana temporaria*, *Rana arvalis*, and *Pleurodeles waltii*;^[2b] 5-*O*-methyl-KDN (KDN5Me) (the sialic acid form in **30**) has been observed in *Sinorhizobium fredii*;^[25] *N*-acetylglycolyl-neuraminic acid (Neu5GcAc) (the sialic acid form in **31**) has been detected in rat peritoneal macrophages.^[26]

Naturally occurring α -2,6-linked sialosides containing a C9-substituted Neu5Ac or KDN (33–35) were synthesized in 72–84% yields (Table 1, entries h-j) from the corresponding C6-modified ManNAc or mannose (8-10). 9-O-Acetylation has been found in nearly all higher animals and certain bacteria and is believed to play a pivotal role in modulating various biological processes.^[27] For example, 9-O-acetyl-Nacetylneuraminic acid (Neu5,9Ac₂) (the sialic acid form in 33) is an essential determinant of the cell surface receptors for the influenza C virus, but prevents the recognition of influenza A and B viruses. [28] It is also presented in an antigenic ganglioside found in developing rat embryonic cells. [29] 9-O-Lactyl-Neu5Ac (Neu5Ac9Lt) (the sialic acid form in 34) has been found in bovine submandibular gland glycoprotein^[30] and human RBC membranes.[2a] 9-O-Acetyl-KDN (KDN9Ac) (the sialic acid form in 35) has been found in O. nerka adonis polysialoglycoproteins.[31]

Naturally occurring α -2,6-linked sialosides containing disubstituted sialic acid residues **36** and **37** have also been synthesized successfully in 78% and 65% yields (Table 1, entries k and 1), respectively. 9-O-Acetyl-Neu5Gc (Neu9-

Scheme 1. An efficient chemoenzymatic approach for the synthesis of sialosides containing sialic acid modifications. PPi=inorganic pyrophosphate, SiaT=sialyltransferase.

Table 1: Synthesis of α -2,6-linked sialosides using the one-pot, three-enzyme system shown in Scheme 1. [a]

| Entry | Donor Precursors | Acceptor | Product | Yield [% |
|-------|--------------------------------|---|---|----------|
| a | HO NHAC HO OH | HO OH HO OH OOH OH HO OOH OOH OON OOH | HO OH CO2 OH ACHN O OH OO OH OOH OOH OOH OOH OOH OOH OOH | 98 |
| Ь | HO OH HO 10 HO 2 | 20 | HO OH CO2 OH HO OH OH OH OH OH OH OH | 97 |
| С | HO HN OH HO JO OH | 20 | HO OH CO2 OH HO HO HO OH HO OH OH OOH OOH OH OH OOH OH OH OOH OH OOH OH OH OH OOH OH | 95 |
| d | HO OAC HO OH | 20 | HO 2HO OH OH OH OH OH | 75 |
| e | HO OMe HO OH | 20 | HO OH CO ₂ OH OH OH OH OH OH OH OH | 76 |
| f | HO HN OAC | 20 | ACO NH HO OH OO OH OOH OOH OOH OOH OOH OOH | 87 |
| g | HO HN OME HO OH 7 | 20 | MeO NH OO OHO OHO OHO OH | 99 |
| h | AcO NHAC HO OH HO 8 | 20 | Achn HO HO OH OH OH OH OH OH | 84 |
| i | HO NHAC HO O OH | 20 | OH OHOH CO2 OH O ACHN OO OHOHO OH O OHOO OH O OHOO OH O OHOO OH | 72 |
| j | AcO OH HO 10 OH | 20 | ACO OH OH CO2 OH OH OH OH OH OH OH | 75 |
| k | AcO HN OH HO OH | 20 | ACO OHOH CO ₂ OH HO NHO OHOO OH OHOO OHOO OHOO OHOO OHOO | 78 |
| I | AcO OAC HO IO OH | 20 | ACO OH OH OH OH OH OH OH | 65 |
| m | OH HO 10 13 | 20 | HO OH CO ₂ OH OH OH OH OH OH OH OH OH | 92 |
| n | HO HN NHCbz | 20 | CbzHN OHO OHO OHO OHO OH | 99 |
| 0 | N ₃ NHAc HO O OH | HO OH HO OMe OH | N ₃ OH CO ₂ OH AcHN HO HO OMe | 93 |

Table 1: (Continued)

| Entry | Donor Precursors | Acceptor | Product | Yield [%] |
|-------|----------------------------------|---|---|-----------|
| р | N ₃ OH HO OH 16 | 21 | N ₃ OH _{OH} CO ₂ OH HO HO OMe | 91 |
| q | HO HN N ₃ HO OH | 21 | $ \begin{array}{c ccccccccccccccccccccccccccccccccccc$ | 90 |
| r | HO N ₃ HO 10 OH | 21 | HO OHOH CO2 OH N3 HO HO OH | 86 |
| S | HO HN OH HO 19 | 21 | HO OH OH CO2 OH O NH OO OH OH OH | 87 |
| t | HO NHAC HO O WOH | HO OH 22 HO NHAC N ₃ | HO OH OH CO2 OH ACHN HO HO NHAC | 87 |
| u | HO NHAC HO O OH | HO OH HO ACHINO 23 ACHINO N ₃ | Achn HO Achn N ₃ | 61 |
| v | HO NHAC HO OH | HO OH HO OH OH NHAC | HO OHOH CO2 OH ACHN OO OHO OH HO HO OH HO NHAC | 92 |
| w | HO NHAC HO OH | HO OH ODO O OH | HO QHOH CO2 ACHN O OHOHO OH HO OHOHO OHOHO OH ACHN OHOHO OHO HO 48 OH HO OH | 91 |

[a] All reactions were carried out in preparative (50–150 mg) scales.

Ac5Gc) (the sialic acid form in **36**) has been found in bovine submandibular gland glycoprotein.^[30] 5,9-Di-*O*-acetyl-KDN (KDN5,9Ac₂) (the sialic acid form in **37**) has also been found in nature.^[1a]

By using the same 3-azidopropyl lactoside **20** as the acceptor for Pd2,6ST, non-natural sialosides containing a 4,6-bis-*epi*-KDO (**38**) and an *N*-(benzyloxycarboxyamido) glycinylamido-neuraminic acid (NeuGlyCbz) (**39**) were successfully synthesized in high yields (Table 1, entries m and n). The successful synthesis of the KDO analogue containing trisaccharide **38** demonstrates that Pd2,6ST can transfer sialic acids with carbon backbones shorter than nine carbons to galactoside with high efficiency. Furthermore, a bulky benzyloxycarbonyl (Cbz) group at C9 of the sialic acid residue of the donor does not affect the activity of Pd2,6ST. This further demonstrates the extremely flexible donor–substrate specificity of this enzyme.

By using GalβOMe (21) as an acceptor for Pd2,6ST, α-2,6-linked sialosides containing sialic acid modified with non-natural substitutents (40–44), such as an azido or an acetylene group, were also synthesized in excellent yields (86–93%) from C2- or C6-modified ManNAc or mannose that bears azide or alkyne functional groups 15–19. The resulting

sialosides tagged with an azide or alkyne group can be easily linked to other molecules through a Staudinger ligation^[32] or click chemistry.^[33]

Acceptor specificity of the recombinant Pd2,6ST was explored by using an *N*-acetyl-β-galactosaminide **22**, an *N*-acetyl-α-galactosaminide **23**, an α-galactoside (methyl-α-D-galactopyranoside), β-galactosides **20**, **21**, and **24**, and an α-2,3-linked sialoside **25**. Similar to that reported previously, [8c,23] Pd2,6ST showed a much broader acceptor specificity compared to mammalian sialyltransferases. [24] The recombinant Pd2,6ST could use both α- or β-linked GalNAc derivatives. Furthermore, β-galactosides and α-2,3-linked sialosides were all excellent acceptors for the sialyltransferase. In contrast to that reported, [23] however, α-galactosides were not acceptors for the enzyme.

In conclusion, we have demonstrated that the recombinant P damsela α -2,6-sialyltransferase does not only have relaxed acceptor–substrate specificity, but also extremely flexible donor–substrate specificity. We have also proved the practical application of the one-pot, three-enzyme chemoenzymatic approach in effective preparative-scale synthesis of diverse sialosides with naturally or non-naturally occurring sialic acid modifications. This system can be easily extended

to large-scale synthesis. Sialyltransferases, together with sialic acid aldolases and CMP-sialic acid synthetases, are important tools for simple and efficient preparation of sialoside libraries. These structurally defined synthetic sialosides will be essential standards for developing analytical methods. They are key probes to elucidate the biological significance of nature's sialic acid diversity, the biosynthetic and degradation pathways of structurally modified sialic acids, and their involvement in the physiological and pathological processes of human and other vertebrates. The synthetic sialosides are also important ligands for identifying specific lectins and producing specific antibodies, which in turn, are vital tools in histochemical studies of organ- and species-specific presentation of modified sialic acids.

Experimental Section

Preparative scale (50–150 mg) synthesis: Reactions were typically carried out in a 50 mL centrifuge tube of Tris-Cl buffer solution (10 mL, 100 mm; pH 8.5 or pH 7.5) containing an acceptor (lactose, GalNAc, or galactose derivatives, 50–100 mg), ManNAc or mannose derivatives (1.5 equiv), sodium pyruvate (7.5 equiv), cytidine-5′-triphosphate (CTP; 1.5 equiv), MgCl₂ (20 mm), aldolase, NmCSS, and Pd2,6ST. The reaction mixture was incubated at 25 °C for 12 h with shaking (120 rpm). After adding the same volume of ice-cold methanol to stop the reaction, insoluble material was removed by centrifugation. The supernatant was concentrated by rotor evaporation and purified by Bio-Gel P2 gel filtration chromatography. Lyophilized sialoside products were characterized by NMR spectroscopy and HRMS.

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- [1] a) T. Angata, A. Varki, *Chem. Rev.* 2002, 102, 439-469; b) R. Schauer, *Glycoconjugate J.* 2000, 17, 485-499.
- [2] a) C. Robbe, C. Capon, E. Maes, M. Rousset, A. Zweibaum, J. P. Zanetta, J. C. Michalski, *J. Biol. Chem.* 2003, 278, 46337 46348;
 b) J. P. Zanetta, A. Pons, M. Iwersen, C. Mariller, Y. Leroy, P. Timmerman, R. Schauer, *Glycobiology* 2001, 11, 663 676.
- [3] M. D. Chappell, R. L. Halcomb, J. Am. Chem. Soc. 1997, 119, 3393–3394.
- [4] a) G. J. Boons, A. V. Demchenko, *Chem. Rev.* **2000**, *100*, 4539 4566; b) M. J. Kiefel, M. von Itzstein, *Chem. Rev.* **2002**, *102*, 471 490.
- [5] a) M. Izumi, C. H. Wong, Trends Glycosci. Glycotechnol. 2001, 13, 345; b) J. Thiem, W. Treder, Angew. Chem. 1986, 98, 1100–1101; Angew. Chem. Int. Ed. Engl. 1986, 25, 1096–1097; c) D. H. Joziasse, W. E. Schiphorst, D. H. Van den Eijnden, J. A. Van Kuik, H. Van Halbeek, J. F. Vliegenthart, J. Biol. Chem. 1987, 262, 2025–2033; d) C. Unverzagt, S. Kelm, J. C. Paulson, Carbohydr. Res. 1994, 251, 285–301; e) M. Gilbert, A.-M. Cunningham, D. C. Watson, A. Martin, J. C. Richards, W. W. Wakarchuk, Eur. J. Biochem. 1997, 249, 187–194.
- [6] a) K. M. Koeller, C. H. Wong, Nature 2001, 409, 232-240;
 b) H. J. Gross, R. Brossmer, Eur. J. Biochem. 1988, 177, 583-589;
 c) H. H. Higa, J. C. Paulson, J. Weinstein, J. Biol. Chem. 1985, 260, 8838-8849;
 d) Y. Kajihara, T. Ebata, K. Koseki, H. Kodama, H. Matsushita, H. Hashimoto, J. Org. Chem. 1995, 60, 5732-5735.

- [7] a) S. J. Danishefsky, J. R. Allen, Angew. Chem. 2000, 112, 882–912; Angew. Chem. Int. Ed. 2000, 39, 836–863; b) C. D. Meo, A. V. Demchenko, G. J. Boons, J. Org. Chem. 2001, 66, 5490–5497; c) V. Martichonok, G. M. Whitesides, J. Org. Chem. 1996, 61, 1702–1706; d) H. Kondo, Y. Ichikawa, C. H. Wong, J. Am. Chem. Soc. 1992, 114, 8748–8750.
- [8] a) O. Blixt, K. Allin, L. Pereira, A. Datta, J. C. Paulson, J. Am. Chem. Soc. 2002, 124, 5739-5746; b) C. H. Wong, R. L. Halcomb, Y. Ichikawa, T. Kajimoto, Angew. Chem. 1995, 107, 569-593; Angew. Chem. Int. Ed. Engl. 1995, 34, 521-546; c) Y. Kajihara, T. Yamamoto, H. Nagae, M. Nakashizuka, T. Sakakibara, I. Terada, J. Org. Chem. 1996, 61, 8632-8635.
- [9] a) Y. Ichikawa, J. L. C. Liu, G. J. Shen, C.-H. Wong, J. Am. Chem. Soc. 1991, 113, 6300-6302; b) E. C. Rodriguez, L. A. Marcaurelle, C. R. Bertozzi, J. Org. Chem. 1998, 63, 7134-7135; c) S. A. Kalovidouris, O. Blixt, A. Nelson, S. Vidal, W. B. Turnbull, J. C. Paulson, J. F. Stoddart, J. Org. Chem. 2003, 68, 8485-8493; d) T. K. Mong, H. K. Lee, S. G. Duron, C. H. Wong, Proc. Natl. Acad. Sci. USA 2003, 100, 797-802.
- [10] a) C. Butor, S. Diaz, A. Varki, J. Biol. Chem. 1993, 268, 10197–10206; b) A. L. Lewis, V. Nizet, A. Varki, Proc. Natl. Acad. Sci. USA 2004, 101, 11123–11128; c) C. Traving, R. Schauer, Cell. Mol. Life Sci. 1998, 54, 1330–1349.
- [11] W. F. Vann, T. Y. Liu, J. B. Robbins, J. Bacteriol. 1978, 133, 1300 1306.
- [12] H. H. Higa, C. Butor, S. Diaz, A. Varki, J. Biol. Chem. 1989, 264, 19427 – 19434.
- [13] V. Vandamme-Feldhaus, R. Schauer, J. Biochem. 1998, 124, 111 121.
- [14] H. H. Higa, A. Varki, J. Biol. Chem. 1988, 263, 8872-8878.
- [15] R. S. Houliston, H. P. Endtz, N. Yuki, J. Li, H. C. Jarrell, M. Koga, A. Van Belkum, M.-F. Karwashki, W. W. Wakarchuk, M. Gilbert, J. Biol. Chem. 2006, 281, 11480–11486.
- [16] M. Iwersen, V. Vandamme-Feldhaus, R. Schauer, Glycoconjugate J. 1998, 15, 895–904.
- [17] J. Tiralongo, H. Schmid, R. Thun, M. Iwersen, R. Schauer, Glycoconjugate J. 2000, 17, 849 – 858.
- [18] A. Kelm, L. Shaw, R. Schauer, G. Reuter, Eur. J. Biochem. 1998, 251, 874–884.
- [19] T. Kawano, S. Koyama, H. Takematsu, Y. Kozutsumi, H. Kawasaki, S. Kawashima, T. Kawasaki, A. Suzuki, J. Biol. Chem. 1995, 270, 16458-16463.
- [20] H. Yu, H. Yu, R. Karpel, X. Chen, Bioorg. Med. Chem. 2004, 12, 6427–6435.
- [21] H. Yu, H. Chokhawala, R. Karpel, H. Yu, B. Wu, J. Zhang, Y. Zhang, Q. Jia, X. Chen, J. Am. Chem. Soc. 2005, 127, 17618–17619.
- [22] T. Yamamoto, M. Nakashizuka, I. Terada, J. Biochem. 1998, 123, 94–100.
- [23] T. Yamamoto, M. Nakashizuka, H. Kodama, Y. Kajihara, I. Terada, J. Biochem. 1996, 120, 104–110.
- [24] a) T. Yamamoto, H. Nagae, Y. Kajihara, I. Terada, *Biosci. Biotechnol. Biochem.* 1998, 62, 210–214; b) T. Endo, S. Koizumi, K. Tabata, S. Kakita, A. Ozaki, *Carbohydr. Res.* 2001, 330, 439–443; c) C.-F. Teo, T.-S. Hwang, P.-H. Chen, C.-H. Hung, H.-S. Gao, L.-S. Chang, C.-H. Lin, *Adv. Synth. Catal.* 2005, 347, 967–972.
- [25] A. M. Gil-Serrano, M. A. Rodriguez-Carvajal, P. Tejero-Mateo, J. L. Espartero, J. Thomas-Oates, J. E. Ruiz-Sainz, A. M. Buendia-Claveria, *Biochem. J.* 1998, 334, 585 – 594.
- [26] A. Kluge, G. Reuter, H. Lee, B. Ruch-Heeger, R. Schauer, *Eur. J. Cell Biol.* **1992**, *59*, 12–20.
- [27] A. Varki, Glycobiology 1992, 2, 25-40.
- [28] a) H. H. Higa, G. N. Rogers, J. C. Paulson, Virology 1985, 144,
 279–282; b) G. Herrler, R. Rott, H. D. Klenk, H. P. Muller,
 A. K. Shukla, R. Schauer, Embo J. 1985, 4, 1503–1506.

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- [29] D. A. Cheresh, R. A. Reisfeld, A. P. Varki, Science 1984, 225, 844–846.
- [30] a) G. Reuter, R. Pfeil, S. Stoll, R. Schauer, J. P. Kamerling, C. Versluis, J. F. Vliegenthart, Eur. J. Biochem. 1983, 134, 139-143;
 b) P. Stehling, M. Gohlke, R. Fitzner, W. Reutter, Glycoconjugate J. 1998, 15, 339-344.
- [31] M. Iwasaki, S. Inoue, F. A. Troy, J. Biol. Chem. 1990, 265, 2596– 2602.
- [32] E. Saxon, C. R. Bertozzi, Science 2000, 287, 2007–2010.
- [33] a) C. W. Tornøe, M. Meldal in *Peptides: The Wave of the Future* (Eds.: M. Lebl, R. A. Houghten), American Peptide Society and Kluwer Academic, San Diego, **2001**, pp. 263–264; b) C. W. Tornøe, C. Christensen, M. Meldal, *J. Org. Chem.* **2002**, 67, 3057–3064; c) H. C. Kolb, M. G. Finn, K. B. Sharpless, *Angew. Chem.* **2001**, 113, 2056–2075; *Angew. Chem. Int. Ed.* **2001**, 40, 2004–2021; d) F. Fazio, M. C. Bryan, O. Blixt, J. C. Paulson, C. H. Wong, *J. Am. Chem. Soc.* **2002**, 124, 14397–14402.